

$$U_g \approx 1 \text{ mV}$$

Temperature compensation for measuring and calibration

Analyte: H2S
 Sensor: 10120117
 Range: 50 mg/l

| Temperature °C | ET |
|----------------|--------|
| 0 | 2,1324 |
| 1 | 2,0453 |
| 2 | 1,9623 |
| 3 | 1,8831 |
| 4 | 1,8077 |
| 5 | 1,7360 |
| 6 | 1,6678 |
| 7 | 1,6029 |
| 8 | 1,5412 |
| 9 | 1,4826 |
| 10 | 1,4270 |
| 11 | 1,3741 |
| 12 | 1,3239 |
| 13 | 1,2762 |
| 14 | 1,2309 |
| 15 | 1,1878 |
| 16 | 1,1468 |
| 17 | 1,1078 |
| 18 | 1,0706 |
| 19 | 1,0350 |
| 20 | 1,0000 |
| 21 | 0,9684 |
| 22 | 0,9371 |
| 23 | 0,9068 |
| 24 | 0,8776 |
| 25 | 0,8491 |
| 26 | 0,8214 |
| 27 | 0,7942 |
| 28 | 0,7675 |
| 29 | 0,7410 |
| 30 | 0,7146 |

Slope:

$$a_{20} = 0,083678 \text{ mg/lmV}$$

H2S concentration:

$$C1 = a_{20} \times (U - U_g) \times ET$$

Total sulphide concentration:

$$C2 = C1 \times G$$

Calibration:

$$a_{20} = aT / ET$$

- C1 - H2S concentration
- a₂₀ - slope at 20°C
- U - U_g - measured voltage - voltage at 0 mg/l H2S
- ET - temperature compensation factor
- aT - slope at calibration temperature
- C2 - total sulphide concentration
- G - total sulphide calculation factor

$$ET = 2,132424E+00 + -8,918868E-02 \times T + 2,097641E-03 \times T^2 + -2,333399E-05 \times T^3 \quad T = [^{\circ}\text{C}]$$